



Semester: 3rd
Programme: B.Pharm

AUTUMN END SEMESTER EXAMINATION-2025
3rd Semester B.Pharm

PHARMACEUTICAL ORGANIC CHEMISTRY-II
BP 301T

(For 2024 Admitted Batch)

Time: 3 Hours

Full Marks: 75

*Answer Question No.1 (Part-A) is compulsory,
any SEVEN from Part-B and any TWO from Part-C.
The figures in the margin indicate full marks.
All parts of a question should be answered at one place only.*

PART-A

1. Objective Answer Type Questions (Answer All) [2 × 10]
- (a) Why is phenol acidic? (2)
 - (b) Define MUFA and PUFA. (2)
 - (c) Mention the steps involved in electrophilic substitution reaction of benzene. (2)
 - (d) Define diazotization reaction. (2)
 - (e) State Huckel's rule for aromaticity. (2)
 - (f) What causes rancidity? (2)
 - (g) Name the oxidised product of anthracene. (2)
 - (h) Mention the theories to explain the stability of cycloalkanes. (2)
 - (i) Why C1-C2 bond length is shorter than C2-C3 in naphthalene? (2)
 - (j) Define polynuclear aromatic hydrocarbons with example. (1+1)

PART-B

2. Focused-Short Answer Type Questions-(Answer Any Seven) [5×7]
- (a) Describe the orbital picture and resonance in benzene with diagrams. (2.5+2.5)
 - (b) Write synthetic uses of diazonium salts. (5)
 - (c) Write the reaction and mechanism of sulphonation of benzene. (2+3)
 - (d) Explain the important reactions of benzoic acid. (5)
 - (e) Discuss the Haworth process for synthesis of anthracene. (5)
 - (f) Give a comparison study between fats and oils. (5)
 - (g) Write a short note on polynuclear hydrocarbons. (5)
 - (h) Classify fatty acids. Explain the reactions of fats and oils. (1+4)
 - (i) Explain the conformations of cyclohexane with energy diagram. (3+2)

PART-C

Long Answer Type Questions (Answer Any Two)

- 3. Explain the preparation and reactions of polynuclear hydrocarbons (naphthalene, anthracene, phenanthrene, diphenylmethane, triphenylmethane). [2+2+2+2] +2]
- 4. Write the structure and medicinal uses of Saccharin, BHC, Chloramine and DDT [2.5+2.5+2.5 +2.5]
- 5. Describe the qualitative tests for phenols. [10]



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AUTUMN END SEMESTER EXAMINATION-2025
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PHYSICAL PHARMACEUTICS-I
BP 302T

(For 2024 Admitted Batch)

Time: 3 Hours

Full Marks: 75

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any SEVEN from Part-B and any TWO from Part-C.
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PART-A

1. Objective Answer Type Questions (Answer All) [2 × 10]
- (a) Define molality. (2)
 - (b) What is Dipole moment? (2)
 - (c) Write down BET equation for Adsorption Isotherm. (2)
 - (d) Define Eutectic mixtures. (2)
 - (e) Describe parachor. (2)
 - (f) Outline Triple point. (2)
 - (g) How refractive index can be determined? (2)
 - (h) List the different terms used with solvent specification to denote solubility. (2)
 - (i) Draw HLB Scale. (2)
 - (j) Highlight the effect of temperature on solubility. (2)

PART-B

2. Focused-Short Answer Type Questions-(Answer Any Seven) [5 × 7]
- (a) Explain in detail about Langmuir adsorption isotherm. (5)
 - (b) Write a note on spreading coefficient. (5)
 - (c) Write a short note on CST. (5)
 - (d) Depict the kinetics of protein binding. (5)
 - (e) Write down the application of buffers in pharmaceutical and other systems. (5)
 - (f) Classify complexation and write its applications. (3+2)
 - (g) Present eutectic mixtures in brief. (5)
 - (h) Depict Nernst distribution law and highlight its limitations. (2+3)
 - (i) Give a brief note on buffered isotonic solutions. (5)

PART-C

Long Answer Type Questions (Answer Any Two)

3. What are the different methods used for measurement of surface and interfacial tension? Write in detail about two methods. [2+4+4]
4. Write down the various diffusion principles in biological systems and explain in brief about gastrointestinal and percutaneous absorption. [2+6+2]
5. Derive Raoult's Law and explain its mathematical expression. Highlight its applications in Ideal and real solutions with deviations. [4+6]



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AUTUMN END SEMESTER EXAMINATION-2025
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PHARMACEUTICAL MICROBIOLOGY
BP 303T

(For 2024 Admitted Batch)

Time: 3 Hours

Full Marks: 75

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any SEVEN from Part-B and any TWO from Part-C.
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PART-A

1. Objective Answer Type Questions (Answer All) [2 × 10]
 - (a) Why are disinfectants not suitable for sterilizing surgical instruments? (2)
 - (b) What is Tyndallisation? Name the scientist associated with this method. (1+1)
 - (c) Differentiate between antiseptics and disinfectants. (1+1)
 - (d) Define phenol coefficient. (2)
 - (e) What are surface active agents? Give examples. (1+1)
 - (f) Define minimum inhibitory concentration (MIC). (2)
 - (g) What is AHVAC system? (2)
 - (h) Why do Gram-negative bacteria not retain crystal violet stain? (2)
 - (i) Why is oxygen toxic to obligate anaerobic bacteria? (2)
 - (j) What is meant by thermal death time in sterilization? (2)

PART-B

2. Focused-Short Answer Type Questions-(Answer Any Seven) [5× 7]
- (a) Describe the principles, types, and applications of Scanning electron microscopy in detail. (3+1+1)
 - (b) Explain the role of preservatives in pharmaceutical preparations and Give the factors affecting Spoilage. (1+4)
 - (c) Outline the principle and method of the staining technique applied in the identification of Mycobacterium. (3+2)
 - (d) Describe the tube dilution method principle, procedure, and advantages. (2+2+1)
 - (e) What is bacteriostatic activity? Discuss the principles and procedures of cup plate. (1+2+2)
 - (f) Describe the different phases of the bacterial growth curve with suitable explanation. (5)
 - (g) Describe clean areas, their classification, and their role in contamination control in pharmaceutical production. (1+3+1)
 - (h) Differentiate between moist heat sterilization and dry heat sterilization with suitable examples. (5)
 - (i) Outline the reproductive mechanisms in fungi, highlighting major types. (5)

PART-C

Long Answer Type Questions (Answer Any Two)

3. Explain in detail the construction and working principle of Laminar Airflow. Add a note on its types, applications, merits, and demerits. [3+3+2
+1+1]
4. Discuss in detail the structure and role of the bacterial cell wall, supported by a labelled diagram. [5+3+2]
5. Discuss different classes of chemical sterilization methods, their mechanisms, examples, and associated advantages and disadvantages. [2+6+1
+1]



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AUTUMN END SEMESTER EXAMINATION-2025
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PHARMACEUTICAL ENGINEERING
BP 304T

(For 2024 Admitted Batch)

Time: 3 Hours

Full Marks: 75

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any SEVEN from Part-B and any TWO from Part-C.
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PART-A

1. Objective Answer Type Questions (Answer All) [2 × 10]
 - (a) What is Reynold's Number? (2)
 - (b) Distinguish between Evaporation and Distillation. (2)
 - (c) Define Filter aids. Give two examples. (2)
 - (d) What is Comminution? (2)
 - (e) Define Fourier's Law. (2)
 - (f) What is equilibrium moisture content? (2)
 - (g) Define the term Elutriation. (2)
 - (h) Give the limitations of Raoult's Law. (2)
 - (i) Define corrosion. Enlist the types of corrosion. (2)
 - (j) Differentiate between conduction, convection and radiation. (2)

PART-B

2. Focused-Short Answer Type Questions-(Answer Any Seven) [5×7]
- (a) Explain Edge and End Runner mill. (3+2)
 - (b) Explain the principle, construction, working, uses, merit and demerits of multiple effect evaporation. (1+1+1+1+1)
 - (c) Explain theories of Filtration. (5)
 - (d) Describe various mechanisms of Heat Transfer. (5)
 - (e) Enumerate theories, types and prevention of Corrosion. (2+1+2)
 - (f) Give the construction, working and uses of Sigma blend mixer. (2+2+1)
 - (g) Explain perforated and non-perforated centrifuge. (2.5+2.5)
 - (h) Explain the principle, construction, working, uses, merit and demerits of Cyclone separator. (1+1+1+1+1)
 - (i) Discuss about molecular Distillation. (5)

PART-C

Long Answer Type Questions (Answer Any Two)

- 3. Write down mechanisms, laws and factors affecting size reduction. Briefly discuss about Ball mill. [2+2+2+4]
- 4. Write down the objective, application and mechanisms of the drying process, also explain Fluidized Bed Dryer. [2+2+2+4]
- 5. Explain Bernoulli's theorem and highlight how it is used to find the rate of flow of fluids in a horizontal pipe. [6+4]
